

Introduction to the use of GCLS

A. Bouazza

**Monash University
Department of Civil Engineering,
Melbourne, Australia**

What is a Geosynthetic Clay Liner?

**Geosynthetic Clay Liners are manufactured
Hydraulic/gas barriers consisting of clay
bonded to a layer or layers of geosynthetics.
They are relatively thin (5 to 7 mm when dry)**

Geosynthetic Clay Liners

Other Names:

- ❧ **Prefabricated Bentonite Blankets**
- ❧ **Bentonite Blankets, Mats, or Panels**
- ❧ **Clay Blankets, Mats, or Panels**

Geosynthetic Clay Liners (GCLs) was selected in 1991 and stuck because:

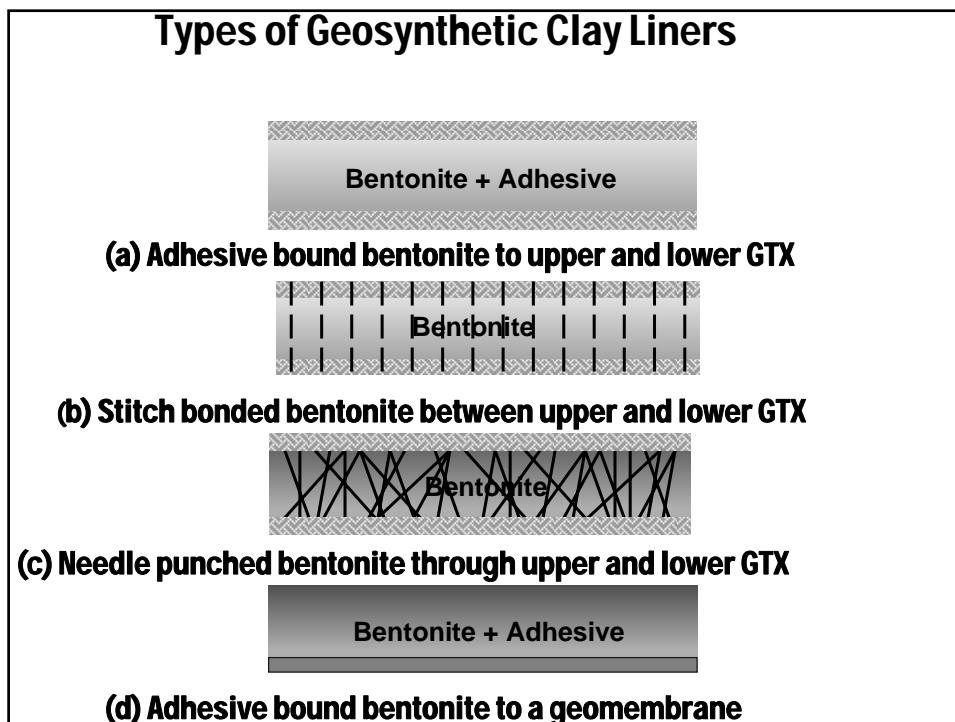
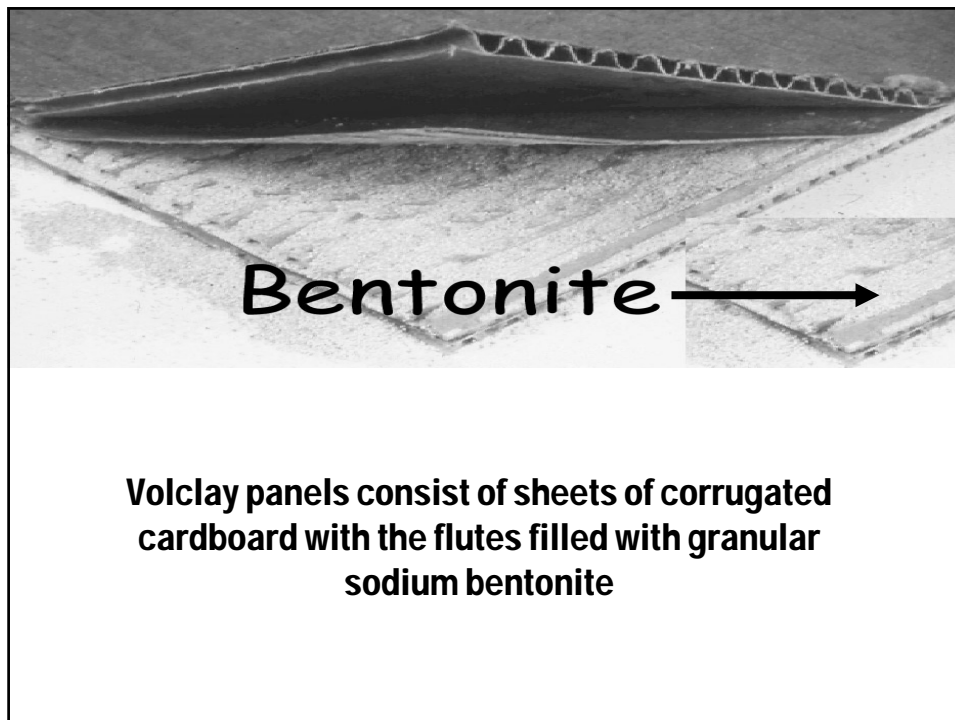
Suggests that GCLs are part of the larger geosynthetic family.

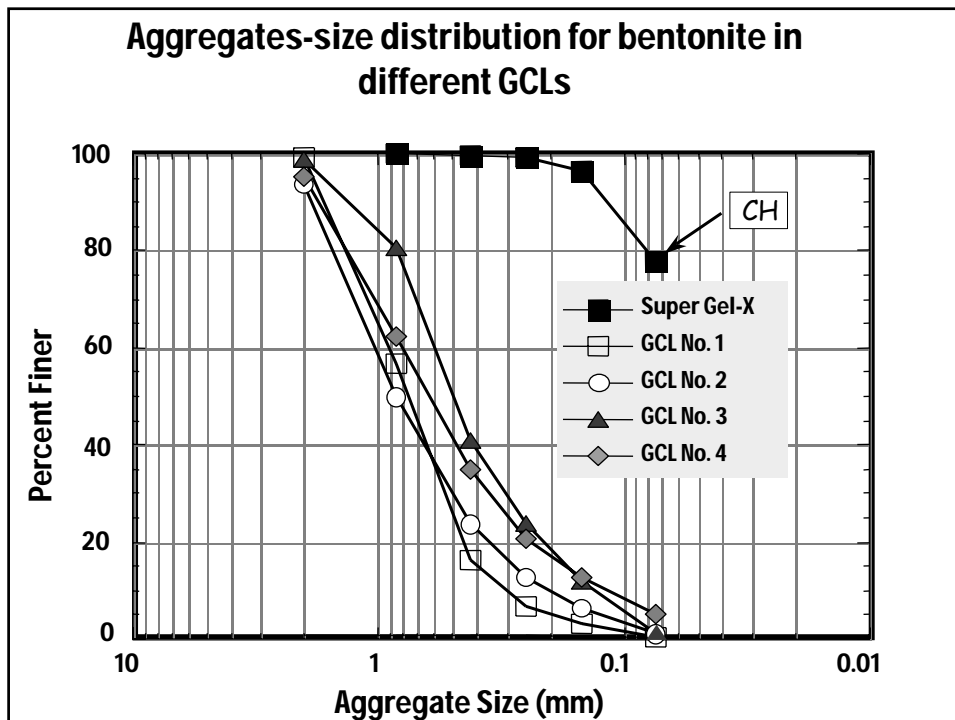
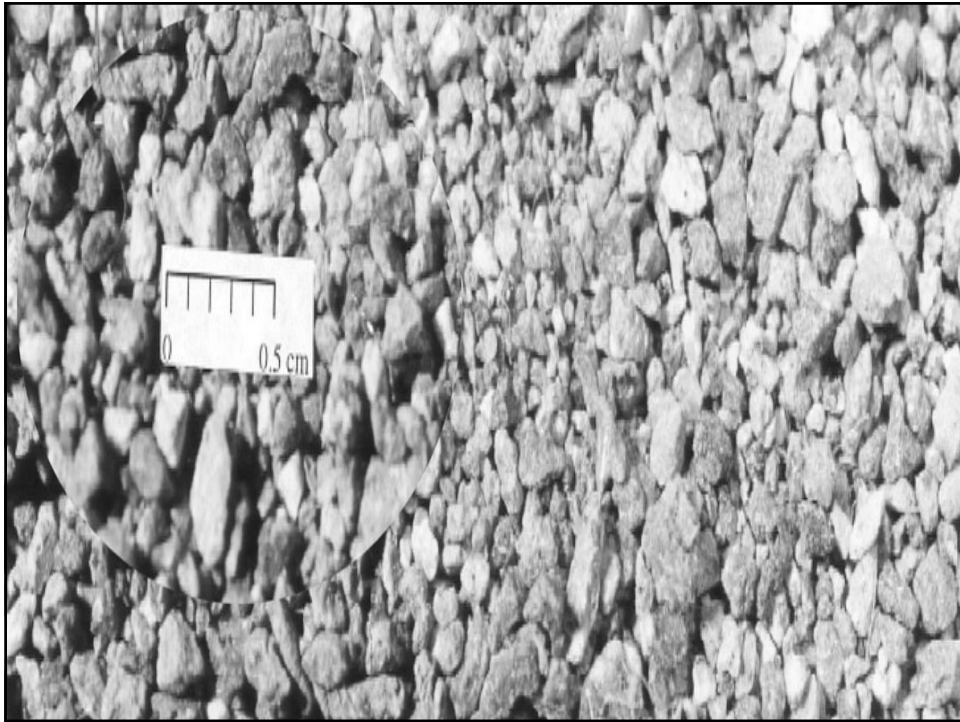
Suggests that GCLs are another type of clay (i.e., an alternative to CCL).

Captures the essence of the material (i.e., Geosynthetic & Clay).

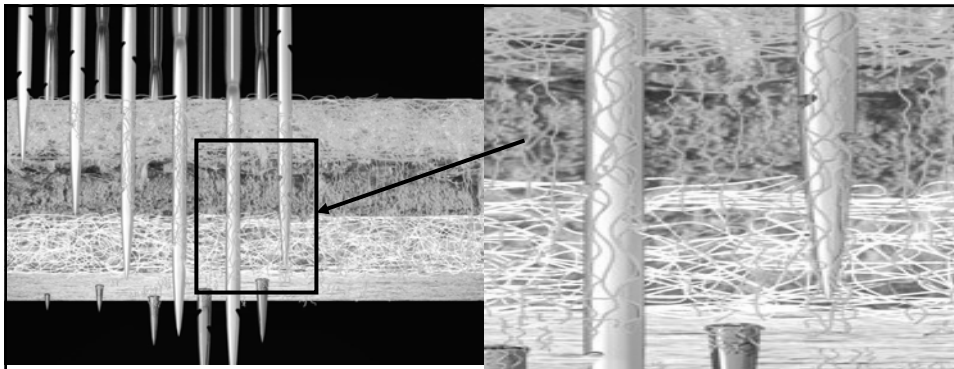
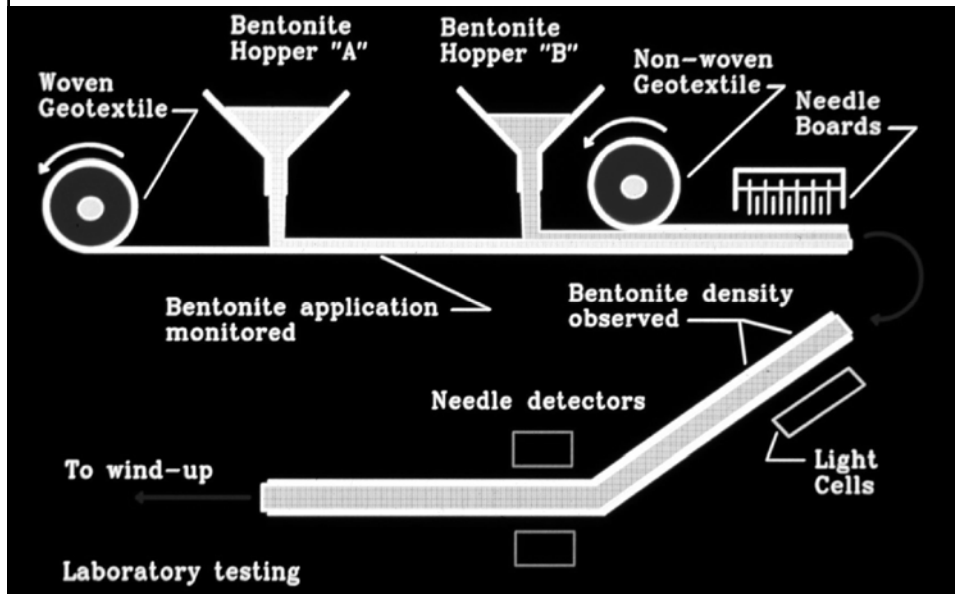
In 1962 Volclay panels were invented and introduced for underground water proof in applications (e.g., basements of commercial buildings)







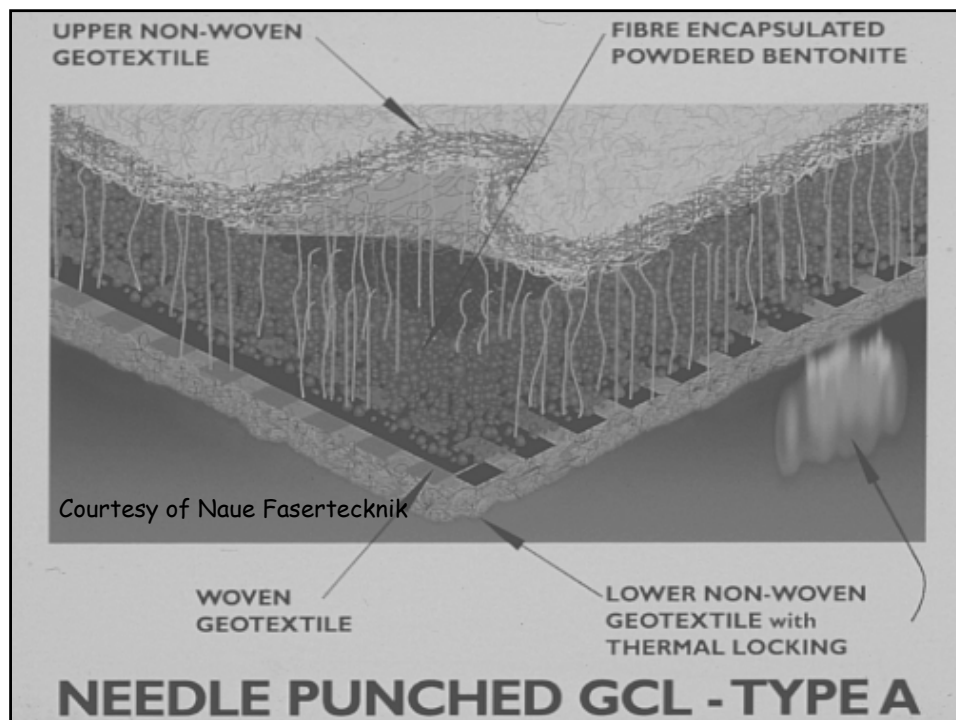
Needle punched Manufacturing Process



Approx. 2.5 million needle-punched fibres per m² create an unique GCL with several important advantages.

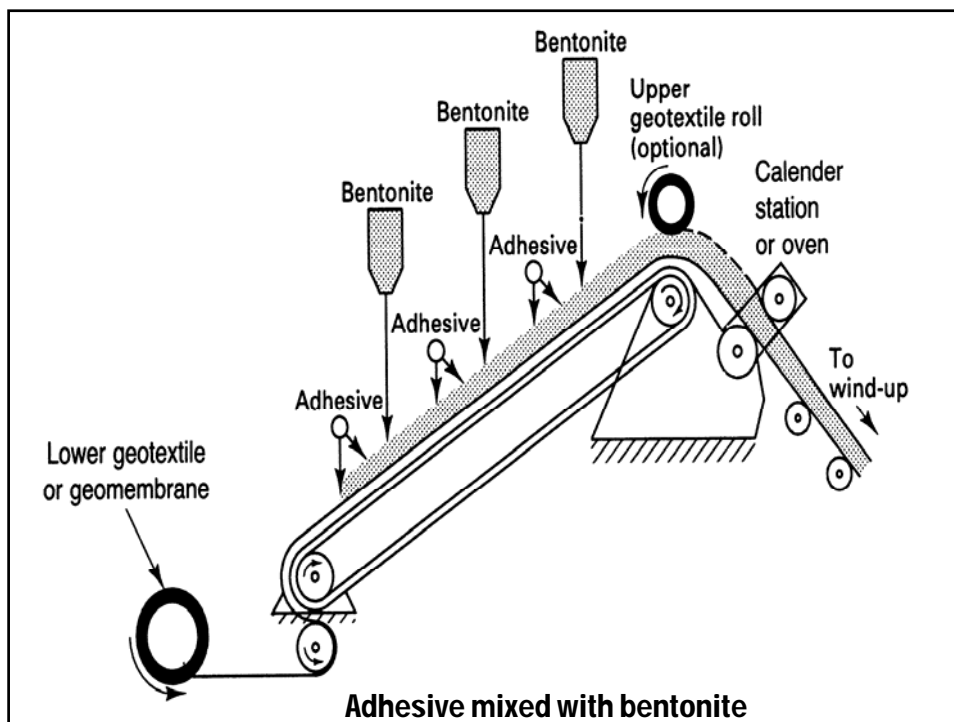
Benefits of Needle punching

- Higher internal shear strength
 - Higher factor of safety
 - Allows steeper slopes or higher shear loads
 - Not the “weakest link” in a multi-layer system
- Better mass per area consistency
 - Fewer thick and thin spots
 - Uniform hydraulic performance
 - Minimal edge loss during handling/installation
- Better resistance to unconfined hydration
 - Resists swelling pressure of bentonite



Benefits of fibre penetration or thermal treatment

- **Larger displacement capability**
 - More shear movement can occur before maximum strength level is achieved
- **Higher post-peak strength**
 - More strength at large displacement
 - Improved FOS in “residual” conditions



Bentonites

Vary considerably in properties due to variations in depositional environment and weathering process.




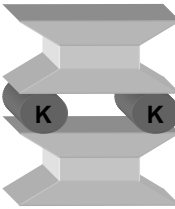

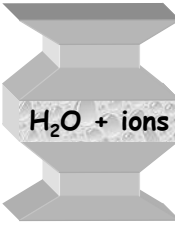
Most bentonites are sodium or calcium bentonites:

- ✓ Volcanic ash that was deposited in marine environments during cretaceous period generally weathered to form sodium (Na^+) bentonite.**
- ✓ Volcanic ash that was deposited in the tertiary period fell mostly in fresh water environments and weathered to form calcium (Ca^{2+}) bentonite**

✓ Purity of bentonite varies from about 60% montmorillonite to 90 % montmorillonite (Other minerals include quartz, feldspar, calcite, and gypsum).

✓ Degree of crushing and pulverisation of bentonite is variable; bentonite granules have different physical characteristics than more finely ground bentonite powder.

✓ Chemicals (particularly polymers) can be added to bentonite ("peptise" the bentonite)

	Structure	Bonding	
Kaolinite			Very stable, with little tendency for volume change when exposed to water
Illite			More plastic than kaolinite. Does not expand when exposed too water unless a deficiency in K exists
Smectite			Because of the weak bond between layers and the negative charge resulting because of isomorphous substitution, the clay absorb water between layers. Has a great tendency for large volume change

Sodium versus Calcium Bentonite

- ✓ Interlayer spacing for Ca^{++} bentonite is 10 to 20 Å; interlayer spacing for Na^+ can theoretically go to infinity but is probably in the range of 20 to 4,000 Å.
- ✓ Due to greater water absorbing capacity in the interlayer zone, Na^+ bentonites tend to have higher water contents, greater shrink-swell potential, and lower hydraulic conductivity

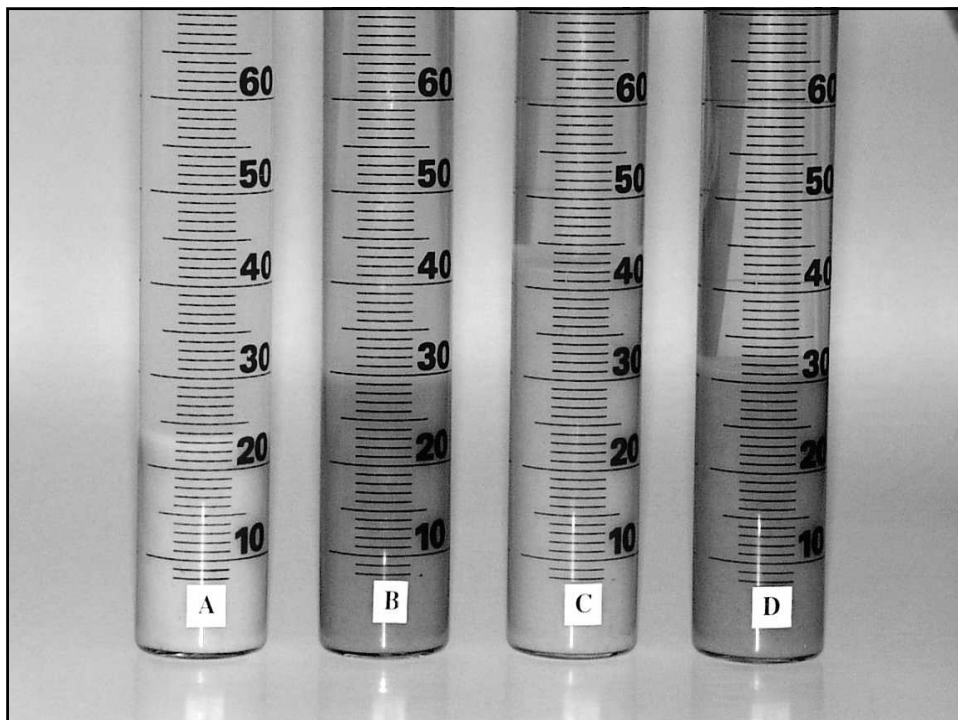
ASTM D5890

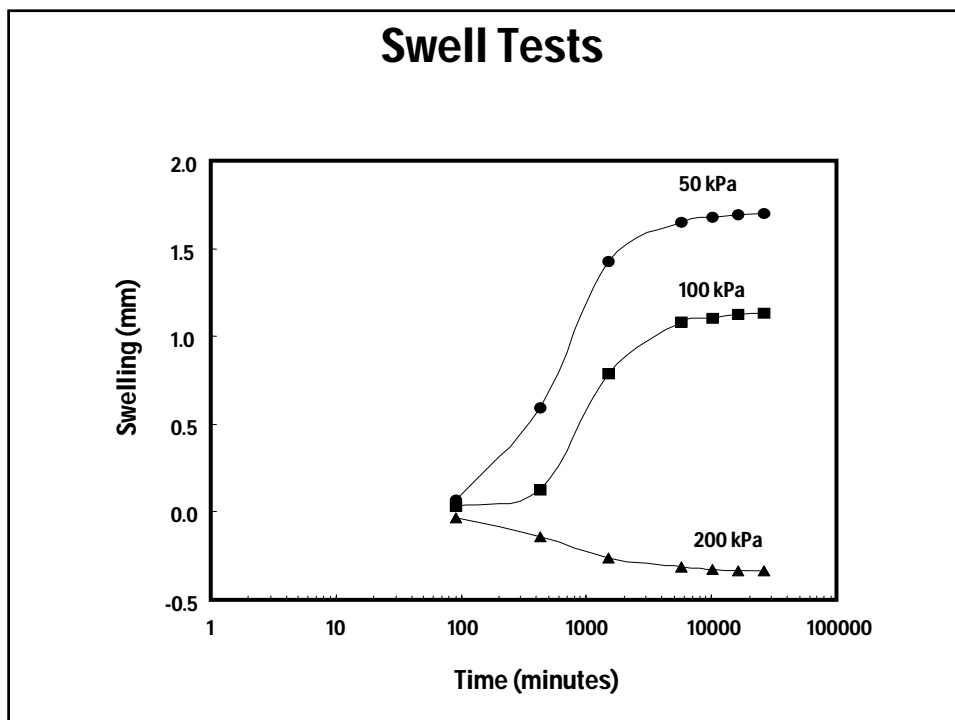
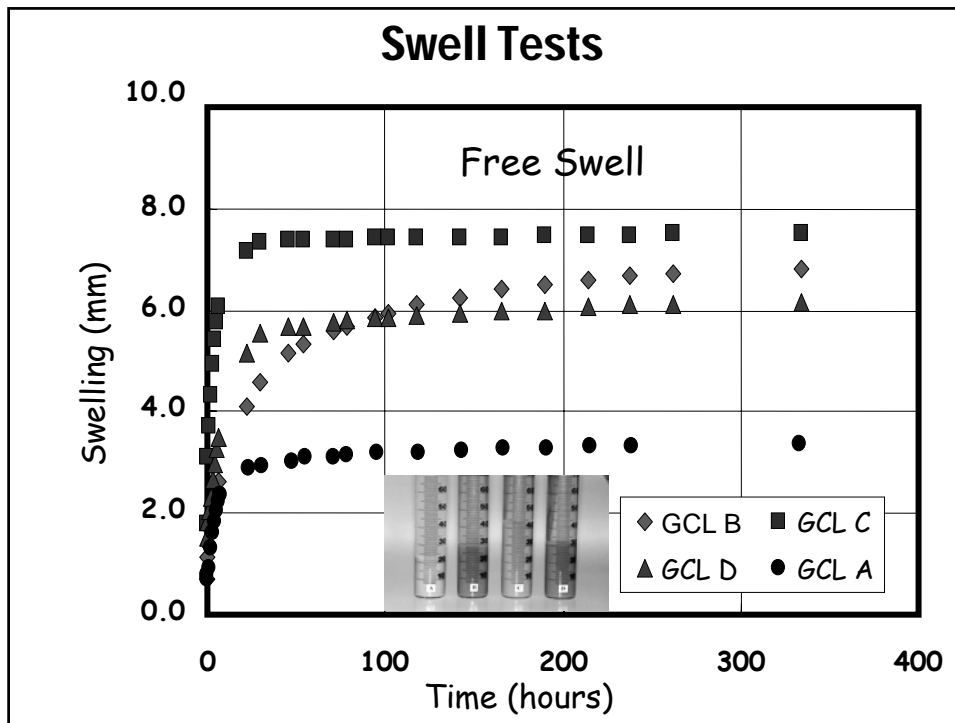
**A graduated cylinder is filled
with 100 ml of water and 2.0 g
of bentonite is added**

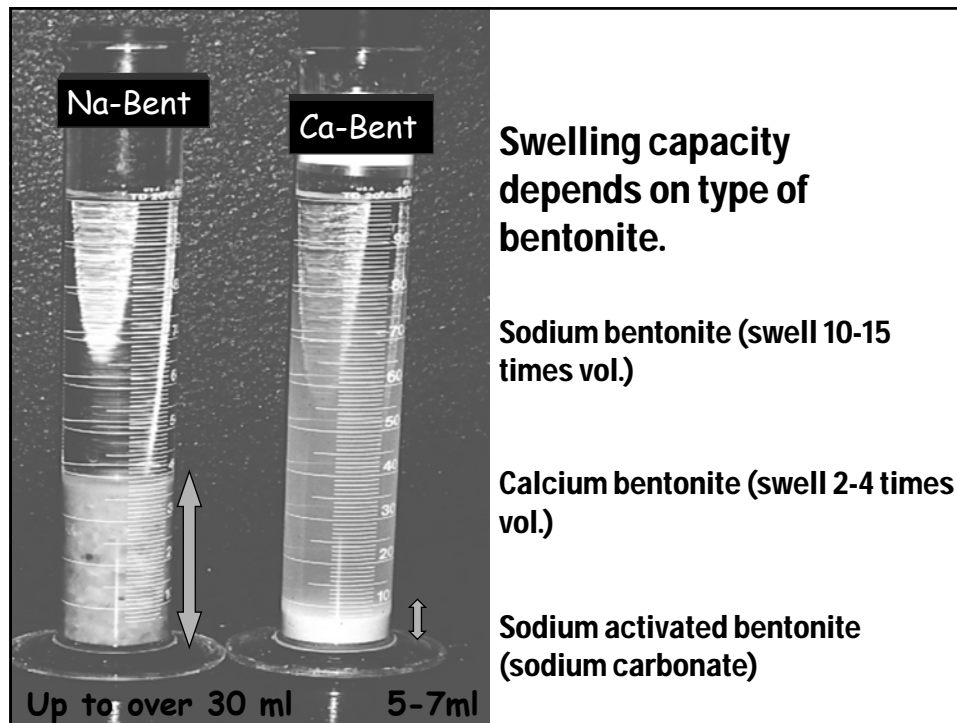
**Volume occupied by the clay
is measured after 24 hrs**



A minimum swell index value of 25 ml is recommended







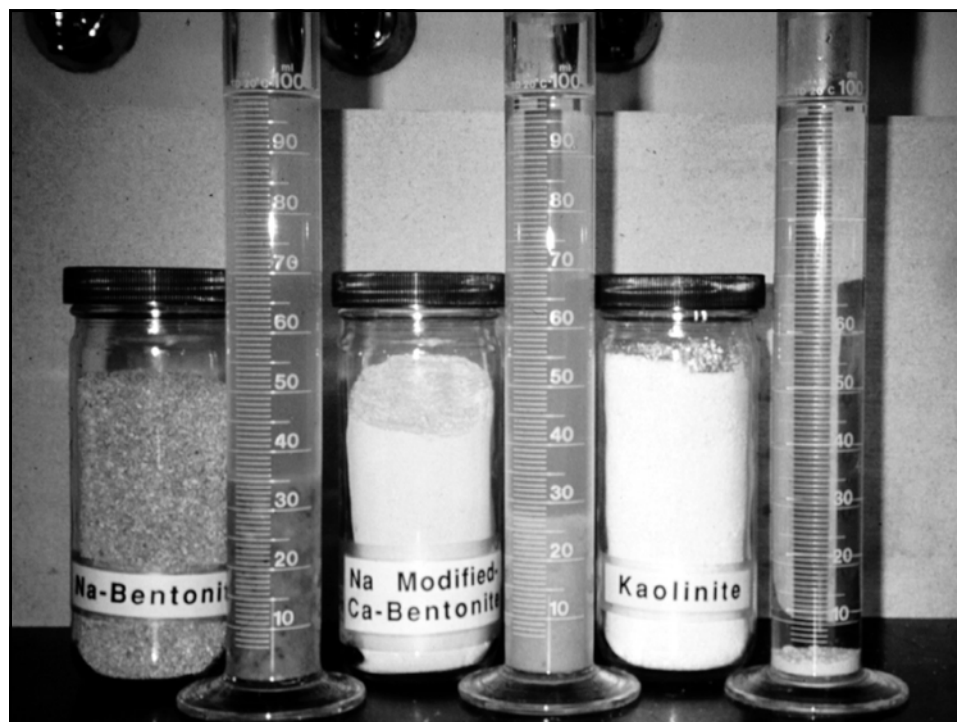
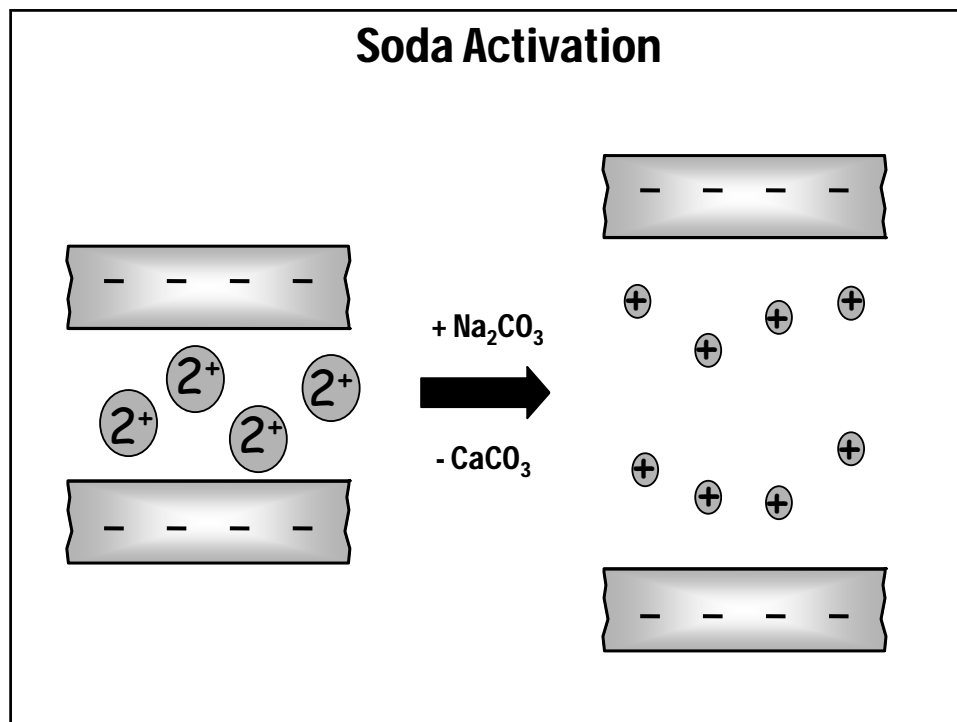
Comparison of Atterberg Limits

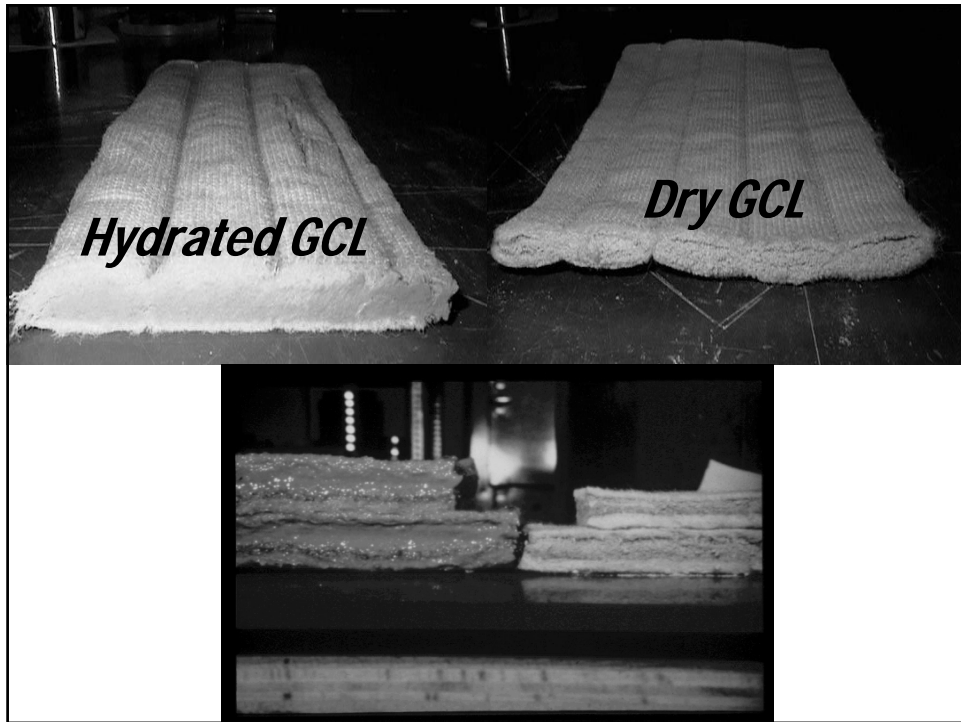
Investigators	LL (%)		PI (%)	
	Na ⁺	Ca ⁺⁺	Na ⁺	Ca ⁺⁺
Grim (1948)	700	124	603	52
Mesri & Olson (1970)	880	200	-	167
Eykholt (1988)	590	123	553	85
Gleason (1993)	603	124	567	98

LL= Water content of the soil when the soil is at the boundary between liquid & plastic state.

PL= Water content of the soil when the soil is at the boundary between plastic and solid state

PI=LL-PI a measure of the water holding capacity over its plastic range





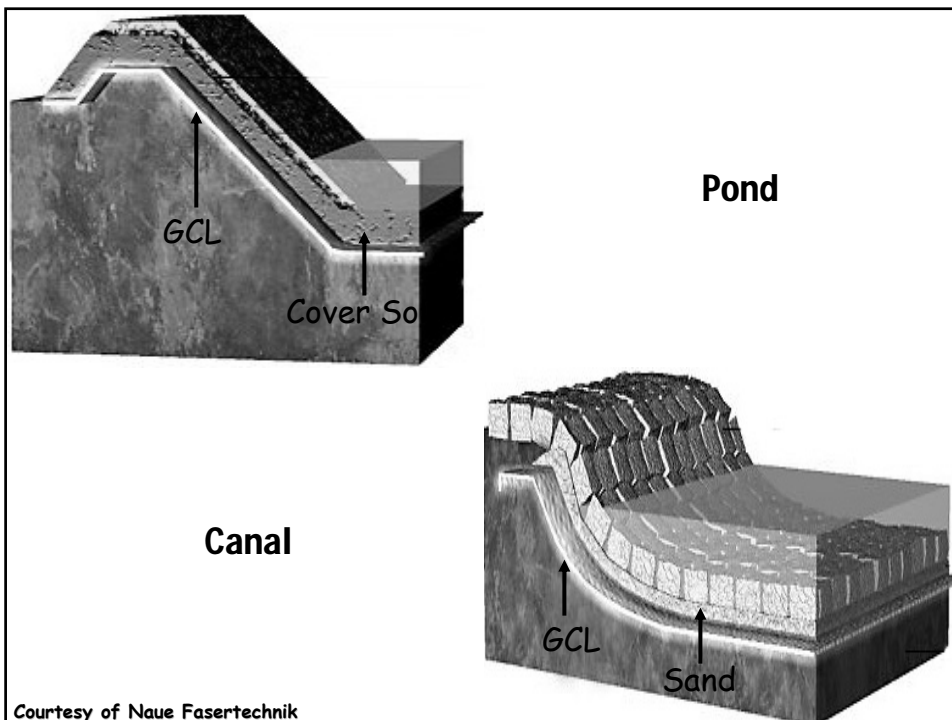
GENERAL APPLICATION OF GCLs

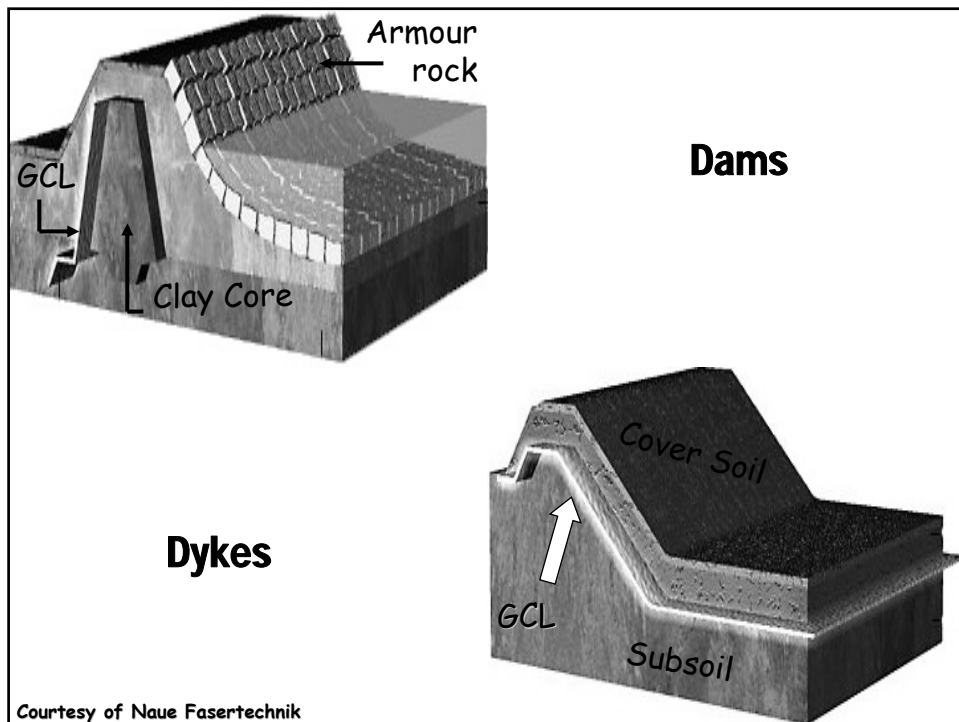
Uses of GCLs in landfill sites

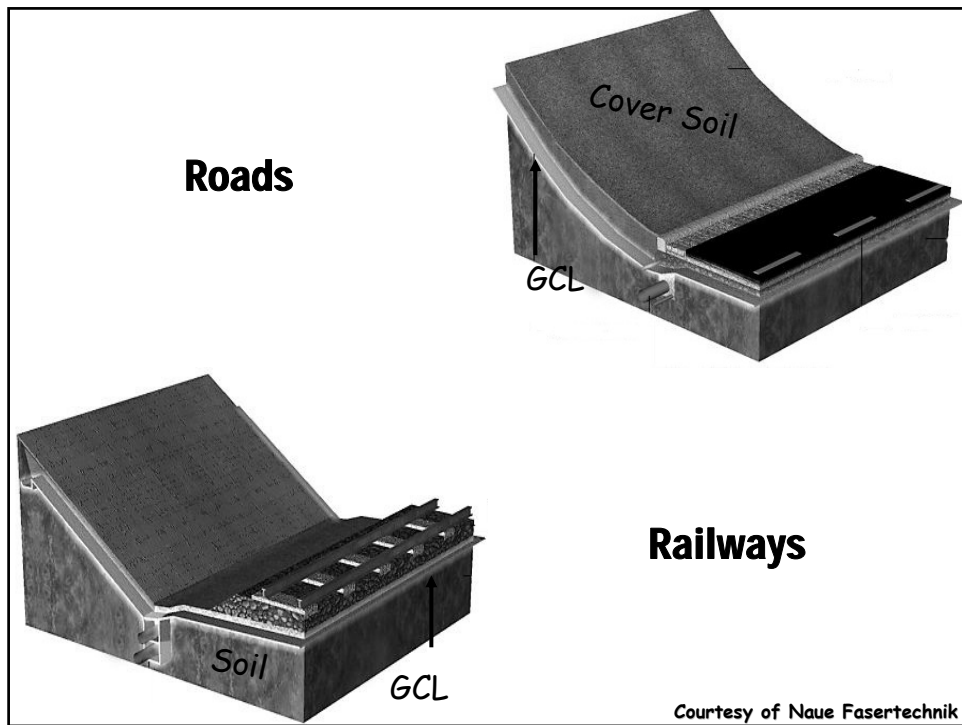


- A liner component
→ Hydraulic barrier

- A cover component
→ Hydraulic barrier
→ Gas barrier







Mining Areas

